

# Chemistry Stoichiometry Problem Sheet 2

## Answer Key

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full credit 1 silver and nitric acid react according  
to the following balanced equation 3 ag s 4 hno<sub>3</sub>  
aq → 3 agno<sub>3</sub> aq 2 h<sub>2</sub>o l a

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web chemistry 51 1 stoichiometry calculation  
practice worksheet calculate the number of  
moles of naoh that are needed to react with 500  
g of h<sub>2</sub>so<sub>4</sub> according to the following equation  
h<sub>2</sub>so<sub>4</sub> 2 naoh na<sub>2</sub>so<sub>4</sub> 2 h<sub>2</sub>o ans 10 mol 2  
calculate the mass of nh<sub>3</sub> that can be produced  
from the reaction of 125 g of ncl<sub>3</sub> according to

the following

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review questions 1 what is avogadro s  
hypothesis 2 how much volume is occupied by  
one mole of a gas at stp 3 in the sample problem  
above assume we combust 1 3 l of propane how  
much co<sub>2</sub> will be produced answers 1 avogadro  
s hypothesis states that equal volumes of all  
gases at

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balance the following chemical equations a hcl o  
2 h<sub>2</sub>o cl<sub>2</sub> b al no<sub>3</sub> 3 naoh al oh<sub>3</sub> nano<sub>3</sub> c h<sub>2</sub>  
n<sub>2</sub> nh<sub>3</sub> d pcl<sub>5</sub> h<sub>2</sub>o h<sub>3</sub>po<sub>4</sub> hcl e fe h<sub>2</sub>so<sub>4</sub> fe  
2 so<sub>4</sub> 3 h<sub>2</sub>fcacl<sub>2</sub> hno<sub>3</sub> ca no<sub>3</sub> 2 hcl g ko<sub>2</sub> h  
2 o koh o<sub>2</sub> h<sub>2</sub>o<sub>2</sub>

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web stoichiometry worksheet and key 1 mol  $\text{KClO}_3$  3 mol  $\text{KClO}_3$  mol  $\text{O}_2$  mol  $\text{O}_2$  3 mol  $\text{KCl}$  mol  $\text{KClO}_3$  0 mol  $\text{Fe}$  mol  $\text{Fe}_2\text{O}_3$  2 mol  $\text{KClO}_3$  è 2 mol  $\text{KCl}$  3 mol  $\text{O}_2$  how many grams of  $\text{O}_2$  will be formed from 3 grams of  $\text{KClO}_3$  how many grams of  $\text{KClO}_3$  are needed to make 30 grams of  $\text{KCl}$  how many grams of  $\text{KCl}$  will be formed from 2 g of  $\text{KClO}_3$

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web key chemistry stoichiometry problem sheet 2 directions solve each of the following problems show your work including proper units to earn full credit 1  $\text{CaCl}_2$  2  $\text{AgNO}_3$  3  $\text{CaNO}_3$  2  $\text{AgCl}$

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web key chemistry stoichiometry problem sheet 1 directions solve each of the following problems show your work including proper units to earn full credit 1 silver and nitric acid react according to the following balanced equation  $3\text{Ag} + 4\text{HNO}_3 \rightarrow 3\text{AgNO}_3 + 2\text{H}_2\text{O} + \text{N}_2\text{O}$

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web answer 355 3g na<sub>2</sub>so<sub>4</sub> 3 using the following equation pb so<sub>4</sub> 2 4 lino<sub>3</sub> pb no<sub>3</sub> 4 2 li<sub>2</sub>so<sub>4</sub> how many grams of lithium nitrate will be needed to make 250 grams of lithium sulfate assuming that you have an adequate amount of lead iv sulfate to do the reaction answer 386 3g of lino<sub>3</sub> 4 using the following equation

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each of the following problems show your work including proper units to earn full credit 1 cacl<sub>2</sub> 2 agno<sub>3</sub> ca no<sub>3</sub> 2 2 agcl how many moles of agcl silver chloride are produced when 3 5 mol of cacl<sub>2</sub> calcium chloride react 3 00 mol cacl<sub>2</sub> x 2 mol agcl 6 00 mol agcl

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$\text{KClO}_3$  molar ratio is 2 : 3 : 2 mol  $\text{KClO}_3$  : 3 mol O  
2 : 12.00 mol  $\text{KClO}_3$  x 18.00 mol x 18.00 mol of O : 2  
3 given the following equation  $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$  how  
many grams of KCl is produced from 250 g of K  
and excess Cl<sub>2</sub>

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